

ENZYMIC ALANYLATION OF AN ISOXAZOLINONE GLUCOSIDE BY LEGUME SEEDLING EXTRACTS

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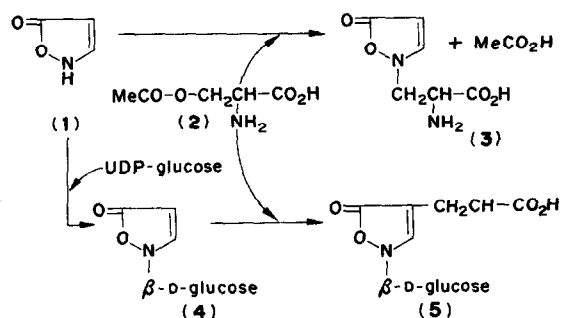
Abstract—Enzymic synthesis of the natural product β -(2- β -D-glucopyranosyl-3-isoxazolin-5-on-4-yl) alanine is described, using the natural isoxazolinone glucoside and *O*-acetyl-L-serine as substrates and extracts from seedlings as enzyme preparations. *Lathyrus odoratus* extracts show a higher activity than those of *Pisum sativum*, *Citrullus vulgaris* and *Leucaena leucocephala*.

INTRODUCTION

Two UV-sensitive heterocyclic β -substituted alanines have been isolated by Lambein *et al.* from pea (*Pisum sativum*) seedlings [1,2]. The structures assigned on the basis of chemical and spectroscopic methods were β -(3-isoxazolin-5-on-2-yl)alanine (3) and β -(2- β -D-glucopyranosyl-3-isoxazolin-5-on-4-yl)alanine (4). Recently, they were isolated also from field pea (*Pisum arvense*), lentil (*Lens culinaris*) and sweet pea (*Lathyrus odoratus*) seedlings [3]. These amino acids are not present in the dry seeds, but they are rapidly formed during the germination.

In recent reports Murakoshi *et al.* [4-6] have described enzymic systems from plants which catalyze the synthesis of heterocyclic β -substituted alanines; serine is first converted to an *O*-acetyl derivative, which undergoes condensation with the appropriate heterocyclic compounds to yield the heterocyclic β -substituted alanines such as mimosine [4], β -pyrazolylalanine [4], quisqualic acid [5], and 3-amino-1,2,4-triazolylalanine [6]. By extension of these observations, Murakoshi *et al.* [7] have reported that crude extracts of *Pisum* seedlings catalyze the synthesis of β -(3-isox-

azolin-5-on-2-yl)alanine (3) from *O*-acetyl-L-serine (2) and 3-isoxazolin-5-one (1) as shown in Scheme 1.



Scheme 1.

We have now further extended these observations and shown that β -(2- β -D-glucopyranosyl-3-isoxazolin-5-on-4-yl)alanine (5) may be synthesized from *O*-acetyl-L-serine (2) and 2- β -D-glucopyranosyl-3-isoxazolin-5-one (4) by an analogous reaction catalyzed by extracts of *Pisum* and *Lathyrus* seedlings as shown in Scheme 1; neither serine nor *O*-phosphoserine could serve as a donor of the alanyl-moiety. The isoxazolinone glucoside

(4) also has been found in *Lathyrus* seedlings by Lambein *et al.* [8] and Van Rompuy *et al.* [9], and synthesized enzymically by Murakoshi *et al.* [10].

RESULTS

Enzyme preparations used in the investigation were crude extracts of *Pisum* and *Lathyrus* seedlings, from which low MW substances were removed by treatment with Sephadex G-25 as described previously [4-7,10]. The enzyme systems and the standard incubation mixtures capable of synthesizing the alanine derivative (5) are described in the Experimental. The reaction product formed in the radioisotopic and non-radioisotopic assays was characterized as 5 by chromatographic comparison with the authentic material. Paper chromatography in two systems showed the presence of a ninhydrin-positive compound with the same R_f as 5 (see Experimental). After ninhydrin spray, both the reaction product and authentic 5 gave a pale yellow brown color that subsequently turned violet. The reaction product was also identified as 5 by TLC on silica gel G by using anthrone- H_2SO_4 spray to visualize carbohydrates. Standard amino acid analysis also confirmed the identity: both the enzymic reaction product and an authentic sample of 5 eluted from the column at a position between taurine and S-methylcysteinesulfoxide.

In the radioisotopic assays *O*-acetyl-L-serine-[3- ^{14}C], was provided as a substrate for the enzymic alanylation, and after incubation, the reaction mixture was analyzed. In the elution profiles of the amino acid analyzer, coincidence was found between peaks of ninhydrin-positive material and of radioactivity emerging at the same position as an authentic sample of 5. Scanning of paper chromatograms also showed incorporation of the [^{14}C]-labeled substrate into 5.

Some properties of the enzyme-dependent synthesis of β -(2- β -D-glucopyranosyl-3-isoxazolin-5-on-4-yl)alanine (5) were studied. Unless otherwise specified, enzyme preparations obtained from *Pisum* seedlings were used as the source of enzyme activity. The rate of synthesis of 5 was constant for at least 120 min but the rate then decreased. The optimum pH for the enzyme-dependent synthesis of 5 was 7.5 (0.1 M potassium

phosphate buffer). The enzyme was active only in a narrow pH range (6.5-8.5).

Since only small amounts of 4 were available, the effect of the concentration of 4 on the formation of 5 was not studied in detail.

The enzyme was reasonably stable. When stored at 0° for 25 hr the remaining activity was about 75% of the activity associated with a freshly prepared extract. Addition of pyridoxal-5'-phosphate up to 25 μ g/ml to the reaction mixture caused neither stimulation nor inhibition of 5 formation.

Enzyme preparations from other plant species were examined for their ability to catalyze the formation of 5 from *O*-acetyl-L-serine and the natural isoxazolinone glucoside (4). The specific activity of enzyme preparations from *Lathyrus* seedlings was approximately 1.5-fold greater than the enzyme preparations from *Pisum* seedlings. Enzyme preparations [4] from watermelon (*Citrullis vulgaris*) and *Leucaena leucocephala*, catalyzing the formation of β -pyrazolylalanine and mimosine, also catalyzed the synthesis of 5 as described for *Pisum* and *Lathyrus* seedling extracts. The initial rates of formation of 5 by *Citrullus* and *Leucaena* enzyme preparations were only slightly lower than those obtained with *Pisum* seedling extracts.

For three natural occurring isoxazolinone derivatives the enzymic synthesis has been described starting from the free isoxazolinone ring (1). While this heterocyclic ring could be formed by chemical interaction of hydroxylamine with nucleic acids, [11,12] no biosynthetic pathway has been reported.

The question whether or not the alanylation at the N-2 position of the free isoxazolinone ring [7] and the alanylation at the C-4 position of the isoxazolinone glucoside involves the same enzyme system is currently under investigation.

EXPERIMENTAL

Labeled chemicals. *O*-acetyl-L-serine-[3- ^{14}C] was synthesized from L-serine-[3- ^{14}C] [13]. The product was diluted with unlabeled *O*-acetyl-L-serine to obtain the sp. act. as used in the different incubation mixtures.

Plant materials. Pea (*Pisum sativum*), watermelon (*Citrullus vulgaris*) and *Leucaena leucocephala* seedlings had grown in moistened vermiculite in the dark for 3-4 days at 30°, and sweet pea (*Lathyrus odoratus*) seedlings for 5-6 days at 30°. After harvest, the testas were removed and the seedlings were cooled at 0° for 30 min before extraction.

Enzyme preparations. All operations were carried out at about 0°. Seedlings were macerated in 0.2 M K-Pi buffer, pH 7.5, containing 0.5% (w/v) 2-mercaptoethanol (3 ml/10 g seedlings). The extract was pressed through fine muslin and centrifuged at 25000 g for 30 min. The supernatant was applied to a column of Sephadex G-25 (fine) equilibrated with 0.1 M K-Pi buffer, pH 7.5, and the same buffer was used to elute the protein fraction and to remove the low MW substances as described in previous papers [4-7,10]. The eluted protein fraction was used immediately as the source of enzyme activity.

Reaction mixtures. The normal reaction mixtures contained 2-β-D-glucopyranosyl-3-isoxazolin-5-one (10 μmol), O-acetyl-L-serine or O-acetyl-L-serine-[3-¹⁴C] (5 μmol, 0.5 μCi) and 0.2 ml of enzyme preparation in a final volume of 0.4 ml. The pH of the incubation mixtures was normally adjusted to pH 7.5 with 0.1 M K-Pi buffer. The mixtures were incubated at 30° and the reaction was terminated by addition of 3 vol. EtOH. The pptd protein was removed by centrifugation, and the supernatant solution was examined chromatographically for the presence of **5**. Occasionally, O-acetyl-L-serine was replaced by L-serine or O-phosphoserine as a donor of the alanyl moiety. Pyridoxal-5'-phosphate (5-25 μg/ml) was added to certain reaction mixtures.

Assay of β-(2-β-D-glucopyranosyl-3-isoxazolin-5-on-4-yl)-alanine (5**) formation.** The formation of **5** was demonstrated by subjecting the supernatant solutions to chromatography, using ninhydrin and anthrone-H₂SO₄ as chromogenic reagents. The product formed was identical with the authentic material (**5**). Paper chromatographic procedures were carried out in: (1) isoPrOH-HCOOH-H₂O (20:1:5, by vol); (2) isoPrOH-H₂O (7:3, v/v). *R_f* values for **5**, O-acetyl-L-serine and L-serine in solvent (1) were 0.17, 0.47 and 0.33 respectively and in solvent (2) 0.23, 0.50 and 0.37 respectively. The reaction product was also characterized as **5** by TLC on Si gel G (Merck type 60) with PhOH-H₂O-EtOH (3:1:1, by wt) by using anthrone-H₂SO₄ for carbohydrates. *R_f* values of **5**, **4** and D-glucose were 0.11, 0.49 and 0.27, respectively.

Further confirmation of the identity of the reaction product as **5** was obtained using AAA under standard operating conditions (150 cm column 50°, 0.2 N NaCl buffer, pH 3.25, flow rate 0.5 ml/min), **5** was eluted from the column at about 162 min (83-84 ml), between the peaks due to taurine and S-methylcysteinesulfoxide. Radioactivity associated with each amino acid peak was recorded with a Packard monitoring flow system coupled to a ratemeter. When paper chromatograms were developed in solvent (1), radioactivity associated

with each ninhydrin-positive substance on the chromatograms was determined using a gas-flow 4π radiochromatogram scanner.

Quantitative determinations of **5** was also made using the cadmium-ninhydrin reagent and the general method described by Atfield and Morris [14] as reported in the previous papers [4-7].

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